

IGCSE Chemistry (9-1) Specification 1(d)

(d) The Periodic Table

Students should:

- **1.18** understand how elements are arranged in the Periodic Table:
 - in order of atomic number
 - in groups and periods
- **1.19** understand how to deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table
- **1.20** understand how to use electrical conductivity and the acid-base character of oxides to classify elements as metals or non-metals
- **1.21** identify an element as a metal or a non-metal according to its position in the Periodic Table
- **1.22** understand how the electronic configuration of a main group element is related to its position in the Periodic Table
- **1.23** understand why elements in the same group of the Periodic Table have similar chemical properties
- 1.24 understand why the noble gases (Group 0) do not readily react